

JONES-HAMILTON CO -- SODIUM BISULFATE/CID AA460 -- 7930-00-559-9481

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Product Identification  
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Product ID:SODIUM BISULFATE/CID AA460

MSDS Date:04/07/1986

FSC:7930

NIIN:00-559-9481

MSDS Number: BJZVY

=== Responsible Party ===

Company Name:JONES-HAMILTON CO

Address:ENTERPRISE AT WILLOW

Box:464

City:NEWARK

State:CA

ZIP:94560

Country:US

Info Phone Num:415-797-2471 OR 419-666-9838

Emergency

Phone Num:415-797-2471(CHEMTREC 800-424-9300)

CAGE:93594

=== Contractor Identification ===

Company Name:JONES-HAMILTON CO.

Address:8400 ENTERPRISE DRIVE

Box:City:NEWARK

State:CA

ZIP:94560

Country:US

Phone:510-797-2471/510-792-4500

CAGE:93594

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Composition/Information on Ingredients  
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Ingred Name:SODIUM BISULFATE ANHYDROUS GLOBULAR (SODIUM ACID SULFATE,  
NITRE CAKE)

CAS:7681-38-1

RTECS #:VZ1860000

Fraction by Wt: 94.5%

Other REC Limits:NONE SPECIFIED

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Hazards Identification  
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Routes of Entry: Inhalation:YES Skin:YES Ingestion:YES  
Reports of Carcinogenicity:NTP:NO IARC:NO OSHA:NO  
Health Hazards Acute and Chronic:INCLUDING SIGNIFICANT ROUTES EFFECTS,  
SYMPTOMS OF OVER EXPOSURE & MEDICAL CONDITIONS AGGRAVATED BY  
EXPOSURE.  
Effects of Overexposure:WATERING OF EYES, MILD SKIN BURNS, MILD BURNING  
SENSATION MAY OCCUR TO MUCOUS MEMBRANES & UPPER RESPIRATORY TRACT.  
INHALATION: IRRITANT. SKIN: MILD TO S  
EVERE IRRITANT. INGESTION:  
IRRITANT.

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First Aid Measures  
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First Aid:EYES:FLUSH THOROUGHLY WITH WATER FOR 15 MINUTES, GET PROMPT  
MEDICAL ATTENTION. SKIN:FLUSH WITH WATER FOR 15 MINUTES. IF BURNS  
OCCUR OBTAIN MEDICAL HELP. INHALE:MOVE TO FRESH AIR LOCATION.  
SWALLOW:DRINK LARGE QUANTITIES OF WATER/MILK, FOLLOW WITH MILK OF  
MAGNESIA, BEATEN EGGS OR VEGETABLE OIL. DO NOT INDUCE VOMITING.  
CONTACT PHYSICIAN IMMEDIATELY.

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Fire Fighting Measures  
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Flash Point:NONE  
Extinguishing Media:WATER OR DRY CHEMICAL AS APPROPRIATE FOR  
COMBUSTIBLES IN AREA  
Fire Fighting Procedures:WHEN WATER USED TO EXTING COMBUSTIBLES/PRODUCT  
IS DISSOLVED IN WATER FORMING SULFURIC ACID.USE ACID PROTECTIVE  
EQUIP. ELEVATED TEMPS(+570F)IS REACHED, USE SCBA  
Unusual Fire/Explosion Hazard:PRODUCT READILY DISSOLVES IN WATER TO  
FORM WEAK SULFURIC ACID SOLUTION. NO GASES, TOXIC FUMES ARE EMITTED  
  
FROM REACTION, BUT PRECAUTIONS SHOULD BE FOLLOWED.

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Accidental Release Measures  
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Spill Release Procedures:SMALL:MATERIAL IS A GRANULAR PRODUCT & CAN BE  
SWEEPED UP FROM SURFACES. LARGE:PICK UP AS MUCH AS POSSIBLE W/SHOVEL  
OR OTHER TOOL. NEUTRALIZE BALANCE OF SPILL W/WEAK ALKALINE SOLUTION  
& WASH DOWN TO SEWER IF FEDERAL, STATE & LOCAL REGULATIONS PERMITS.

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Handling and Storage  
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Handling and Storage

Precautions:AVOID CONTACT WITH SKIN, EYES OR CLOTHING. DO NOT STORE WHERE EXPOSED TO MOIST CONDITIONS OR NEAR STRONG ALKALIES. KEEP CONTAINERS TIGHTLY CLOSED.

Other Precautions:WEAR ALL RECOMMENDED PROTECTIVE EQUIPMENT WHEN HANDLING.

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===== Exposure Controls/Personal Protection =====

Respiratory Protection:NIOSH OR MSA CERTIFIED DUST MASK SHOULD BE WORN WHILE HANDLING PRODUCT TO CONTROL EXPOSURE BELOW NUISANCE DUST LIMITS OF 10 MG/M3

Ventilation:LOCAL EXHA

UST:LOCAL VENTILATION TO A DUST COLLECTOR IS RECOMMENDED.

Protective Gloves:ACID RESISTANT GLOVES:RUBBER OR NEOPRENE

Eye Protection:SAFETY GOGGLES OR FULL FACE SHEILD

Other Protective Equipment:CLOTHES SHOULD COMPLETELY COVER SKIN TO AVOID SKIN CONTACT. COATS, COVERALLS OR APRONS ARE RECOMMENDED.

Work Hygienic Practices:USE COMMON SENSE & CARE AROUND CHEMICALS. NEVER MIX CHEMICALS. CONSULT YOUR SUPERVISOR FOR OTHER PRACTICES.

Supplemental Safety and Health

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===== Physical/Chemical Properties =====

HCC:C1

Melt/Freeze Pt:M.P/F.P Text:350F,177C

Solubility in Water:100%

Appearance and Odor:OFF-WHITE, BEAD LIKE, GRANULAR, DRY MATERIAL

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===== Stability and Reactivity Data =====

Stability Indicator/Materials to Avoid:YES

AVOID CONTACT WITH STRONG ALKALINE MATERIAL SUCH AS CAUSTIC. REACT WITH WATER TO FORM WEAK SULFURIC ACID SOLUTION.

Stability Condition to Avoid:STORE IN DRY AREA TO AVOID MOISTURE CONTACT

Hazardous De

composition Products:NONE UNLESS HEATED OVER 570F AT WHICH SULFUR DIOXIDE & SULFUR TRIOXIDE ARE FORMED.

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===== Disposal Considerations =====

Waste Disposal Methods:COMPLY WITH ALL THE STATE, FEDERAL & LOCAL REGULATIONS.

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